

Polymethine dyes — furo[2,3-b]-6-pyridine derivatives — were synthesized. Replacement of the vinylene group in the benzene ring of the quinoline residue by an oxygen atom and by a sulfur atom causes a bathochromic shift of the absorption maximum of the corresponding cyanine dyes.

It has been shown that replacement of the vinylene group by a sulfur atom in the benzene ring of the quinoline residue in cyanine dyes leads to a bathochromic shift of their absorption maxima [1, 2]. To ascertain the effect on the spectra of the dyes of a similar replacement of the vinylene group by an oxygen atom we synthesized polymethine dyes of the cyanine (A) and merocyanine (B) series and p-dimethylaminostyryl dyes (C) containing a thieno- or furo[2,3-b]-6-pyridine residue (I, II) [3].

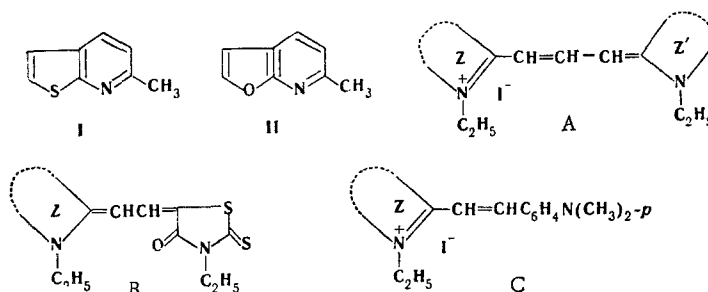


TABLE 1. Absorption Maxima of Carbocyanines (A) in Alcohol

Compound	Z	Z'	λ_{max} , nm	Hypsochromic shift, nm
III	Furo [2,3-b]-6-pyridine	Furo [2,3-b]-6-pyridine	620	—
IV	2-Quinoline	Benzoxazole	545 ^a	1,0
V	Thieno [2,3-b]-6-pyridine	Benzoxazole	542	9,5
VI	Furo [2,3-b]-6-pyridine	Benzothiazole	536	16,5
	2-Quinoline	Benzothiazole	580 ⁵	2,5
VII	Thieno [2,3-b]-6-pyridine	Benzothiazole	572	16,0
	Furo [2,3-b]-6-pyridine	Benzothiazole	571	18,0
VIII	Thieno [2,3-b]-6-pyridine	2-Quinoline	608	4,5
	Furo [2,3-b]-6-pyridine	2-Quinoline	609	4,5
IX	2-Quinoline	4-Quinoline	656	3,0
	Thieno [2,3-b]-6-pyridine	4-Quinoline	662	2,5
	Furo [2,3-b]-6-pyridine	4-Quinoline	663	2,5

*The λ_{max} values in ethanol of symmetrical carbocyanines with benzoxazole, benzothiazole, 2-quinoline, 4-quinoline, and thieno[2,3-b]-6-pyridine residues are, respectively, 485, 558, 607, 711 [5], and 618 nm [1, 2].

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TABLE 2. Absorption Maxima of Dimethinemerocyanines (B) and p-Dimethylaminostyryl Dyes (C) in Alcohol

Compound	Z	λ_{max} , nm	Hypsochromic shift, * nm
B	2-Quinoline	565	9,5
B	Thieno [2,3-b]-6-pyridine	576	4,0
B (X)	Furo [2,3-b]-6-pyridine	574	7,0
C	2-Quinoline	532	76,5
C (XI)	Thieno [2,3-b]-6-pyridine	502	112
C (XII)	Furo [2,3-b]-6-pyridine	488	127,0

* These are the deviations in the λ_{max} values of the dyes from the arithmetic mean values calculated from the λ_{max} values of the corresponding symmetrical carbocyanines and monomethineoxanine derivative of 3-ethylrhodanine (λ_{max} in alcohol 542 nm [6]) or the hydrol of Michler's blue (λ_{max} in nitromethane 610 nm).

TABLE 3. Excitation Energies of Cyanine Dyes

Compound	E_{theor}		E_{exp} , eV	λ_{max} , nm	
	β units	eV		theor.	exptl.
Bis (7-ethylfuro [2,3-b]-6-pyridine) trimethine-cyanine (III)	0,553	1,980	2,00	626	620
1,1'-Diethylquino-2,2'-carbocyanine	0,5717	2,04	2,04	607	607
Bis (7-ethylthieno [2,3-b]-6-pyridine) trimethine-cyanine (XIII)	0,556	1,990	2,006	623	618
3,3'-Diethylthiacarbocyanine	0,6507	2,32	2,21	534	558
Bis (3-ethylthieno [2,3-d]-2-thiazole) trimethine-cyanine (XIV)	0,577	2,062	2,08	600	594

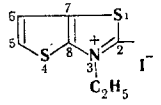
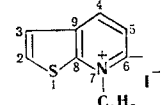
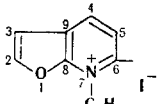
The dyes were obtained by known methods for the synthesis of such compounds (see [4]). The data in Table 1 show that replacement of the vinylene groups by an oxygen atom in the benzene ring of the 2-quinoline residue in the carbocyanines causes a somewhat greater bathochromic shift of the absorption maxima of the dyes than replacement of it by a sulfur atom. A comparison of the magnitudes of the hypsochromic shifts for unsymmetrical carbocyanines shows that the basicities of the furo[2,3-b]-6-pyridine and thieno[2,3-b]-6-pyridine residues are higher than the basicity of the 2-quinoline residue. The basicity of the furo[2,3-b]-6-pyridine residue somewhat exceeds the basicity of the thieno[2,3-b]-6-pyridine residue.

A comparison of the magnitudes of the hypsochromic shifts for p-dimethylaminostyryl dyes (C, Table 2) leads to the same conclusion.

The excitation energies calculated by the molecular orbital method within the Hückel approximation (E_{theor}) [7] and the corresponding maxima on the absorption curves of alcohol solutions of the dyes (E_{exp}) are presented in Table 3. The coulombic and exchange integrals (Table 4) were taken with allowance for the electronegativities of the heteroatoms [8] and the C-X bond lengths (where X is a heteroatom; see [7]). The coulombic integrals of the heteroatoms (α_X) and the carbon atoms bonded to them [$\alpha_{C(X)}$] were calculated from the formulas $\alpha_X = \alpha + h_X\beta$ and $\alpha_{C(X)} = \alpha + 0.1 h_X\beta$ for $h_{-NR-} = +1.5$, $h_{-O-} = 2.0$, and $h_{-S-} = 0.7$; 3.58 eV was adopted for $|\beta|$ [7].

The calculated excitation energies of the electron from the higher occupied level to the lower vacant level increase on passing from the furo[2,3-b]-6-pyridine derivative (dye III) to the thienopyridocarbocyanine and, especially, to the quino-2,2'-carbocyanine; this is in agreement with the observed absorption maxima of these dyes. It has been shown that dyes with thieno[2,3-d]thiazole residues [9] have considerably deeper colors than the corresponding benzothiazole derivatives. These results are also in agreement with the calculated values.

TABLE 4. Exchange Integrals (β_{CC} , β_{CX}) and Energy Levels of Carbocyanine Molecules

Compound	Heterocycle residue	Bond	k_{CC} or k_{CX}^*	Energy levels, β units	
				vacant	occupied
XIV		S ₁ -C ₂	0,4	-1,7765	0,2405
		S ₁ -C ₇		-1,6565	0,5745
		S ₄ -C ₅		-1,6405	0,6565
		S ₄ -C ₈		-1,1915	0,6655
		C ₂ -N ₃	1,0	-0,8695	0,8235
		N ₃ -C ₈		-0,7765	0,9185
				-0,3365	1,0685
					1,4895
					1,6575
					1,7705
		2,6075			
		2,6350			
XIII		N ₇ -C ₈	0,8	-1,9652	0,2548
		N ₇ -C ₆		-1,8762	0,6564
		S ₁ -C ₈	0,6	-1,6842	0,6817
		S ₁ -C ₂		-1,3920	0,7335
			0,9	-1,1282	0,9667
				-0,9375	1,2692
				-0,8248	1,4234
				-0,6711	1,5622
				-0,3008	1,7494
					1,8476
		2,4407			
		2,4736			
III		N ₇ -C ₈	0,7	-1,9740	0,3031
				-1,8673	0,7625
		N ₇ -C ₆	0,8	-1,6701	0,8927
				-1,3964	1,0446
		C ₃ -C ₉	0,9	-1,0876	1,1551
		C ₈ -C ₉		-0,7988	1,2629
		C ₄ -C ₉		-0,6341	1,5678
		C ₅ -C ₆		-0,5377	1,6096
				-0,2502	2,1758
					2,2414
		2,5952			
		2,6069			

* $\beta_{CC} = k_{CC}\beta$; $\beta_{CX} = k_{CX}\beta$.

EXPERIMENTAL METHOD

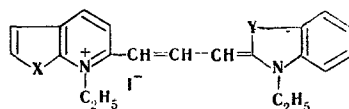
The electronic absorption spectra of ethanol solutions of the dyes were recorded with an SF-2 spectrophotometer.

Bis(7-ethylfuro[2,3-b]-6-pyridine)trimethinecyanine Iodide (III). A 0.58-g (2 mmole) sample of the ethiodide of II was heated for 5-10 min with 0.87 g (2 mmole) of 6- β -acetanilidovinylfuro[2,3-b]pyridine and 0.20 g (1 mmole) of triethylamine in 4 ml of acetic anhydride at 100-105°. The mixture was cooled and diluted with ether, and the resinous precipitate was dissolved by heating in 5 ml of ethanol. A 6-ml sample of 10% potassium iodide solution was added to the ethanol solution, the mixture was cooled, and the precipitated dye was removed by filtration and purified by recrystallization from ethanol to give dark-blue prisms (from ethanol) with mp > 200° (dec.) in 13% yield. Found: C 54.7; H 4.5; I 27.4%. C₂₁H₂₁IN₂O₂. Calculated: C 54.8; H 4.6; I 27.5%.

Unsymmetrical Carbocyanines (IV-IX, Table 5). A 2-mmole sample of the ethiodide of I or II was heated with 2 mmole of the ethiodide of 2- β -acetanilidovinyl-substituted benzoxazole, benzothiazole, or quinoline or the ethiodide of 4- β -acetanilidovinylquinoline and 2 mmole of triethylamine in 5-6 ml of acetic anhydride at 125-130° for 20-30 min, after which the mixture was cooled, and the precipitated dyes were removed by filtration. The products were purified by chromatography on aluminum oxide in chloroform and by recrystallization from ethanol. The dyes crystallized in the form of violet (IV-VII) or green (VIII, IX) prisms.

2-Thioxo-3-ethyl-5-(7-ethylfuro[2,3-b]dihydro-6-pyridylideneethylidene)-4-thiazolidone (X). A mixture of 0.58 g (2 mmole) of the ethiodide of II, 0.60 g (2 mmole) of 2-thioxo-5-

TABLE 5



Compound	x	y	mp, °C	Empirical formula	Found, %			Calc., %			Yield, %
					C	H	I	C	H	I	
IV	S	O	226—227	C ₂₁ H ₂₁ IN ₂ OS	56,7	4,6	28,5	56,7	4,7	28,6	29
V	O	O	232—233	C ₂₁ H ₂₁ IN ₂ O ₂	54,7	4,5	27,5	54,8	4,6	27,5	24
VI	O	S	240—242	C ₂₁ H ₂₁ IN ₂ OS	56,6	4,6	28,5	56,7	4,7	28,6	36
VII	O	CH=CH	256—257	C ₂₃ H ₂₃ IN ₂ O	58,4	4,8	26,9	58,5	4,9	27,0	34
VIII*	S	—	192—194	C ₂₃ H ₂₃ IN ₂ S	56,7	4,7	N 5,7	56,8	4,7	N 5,8	33
IX*	O	—	222—223	C ₂₃ H ₂₃ IN ₂ O	58,4	4,8	26,9	58,5	4,9	27,0	31

* Unsymmetrical carbocyanines with a 4-quinoline residue.

acetanilidomethylene-4-thiazolidone, and 0.25 g (2.5 mmole) of triethylamine in 10 ml of anhydrous ethanol was heated at 95° for 1 h. Workup gave violet prisms (from ethanol) with mp 134–136° in 32% yield. Found: C 57.6; H 4.7%. C₁₆H₁₆N₂O₂S₂. Calculated: C 57.6; H 4.8%.

6-(p-Dimethylaminostyryl)furo[2,3-b]pyridine Ethiodide (XII). A mixture of 0.58 g (2 mmole) of the ethiodide of II, 0.34 g (2 mmole) of p-dimethylaminobenzaldehyde, and 0.09 g (1 mmole) of piperidine in 5 ml of pyridine was heated at 95° for 1 h. The dye was isolated and purified as in the case of the carbocyanines to give brown prisms (from ethanol) with mp 234–236° in 21% yield. Found: C 54.2; H 4.9; I 30.1%. C₁₉H₂₁IN₂O. Calculated: C 54.3; H 5.0; I 30.2%.

6-(p-Dimethylaminostyryl)thieno[2,3-b]pyridine Ethiodide (XI). This compound was similarly obtained as dark-violet prisms (from ethanol) with mp 256–257° in 23% yield. Found: C 52.1; H 4.8; N 6.3%. C₁₉H₂₁IN₂S. Calculated: C 52.3; H 4.8; N 6.4%.

LITERATURE CITED

1. P. I. Abramenko and V. G. Zhiryakov, *Khim. Geterotsikl. Soedin.*, 621 (1967).
2. V. G. Zhiryakov and P. I. Abramenko, *Trudy Gosniikhimfotoproekt*, No. 1, 26 (1968).
3. P. I. Abramenko and V. G. Zhiryakov, *Khim. Geterotsikl. Soedin.*, 1544 (1972).
4. F. M. Hamer, *The Cyanine Dyes and Related Compounds*, New York (1964).
5. N. I. Fisher and F. M. Hamer, *Proc. Roy. Soc. (London)*, 154A, 703 (1936).
6. M. V. Deichmeister, I. I. Levkoev, and É. B. Lifshits, *Zh. Obshch. Khim.*, 23, 1529 (1953).
7. T. Tani and S. Kikuchi, Report of the Institute of Industrial Science, The University of Tokyo, 18, 76 (1968).
8. R. Daudel, R. Lefebvre, and C. Moser, *Quantum Chemistry*, Interscience Publ., New York (1959).
9. V. G. Zhiryakov, Dissertation [in Russian], Moscow (1966).